

Effect of Ti-Doping on Thermo-Optical and Electronic Properties of WS₂ Nanosheets

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ABSTRACT— Ti-doped tungsten disulfide (WS₂) nanosheet-semiconductor is studied for thermo-optical and electronic properties. Thermal diffusivity (D) thermal conductivity (κ) and absorbance were determined as a function of Ti-dopant (0, 7, 14, and 28%). The research focused on the effect of different Ti-dopant concentrations, and we tried to evaluate the thermal parameters using photothermal lens technique as a simple, non-contacting method. The results show an increase in the values of D by 5 times with an increment of Ti-dopant from 0% to 28%. The addition of Ti did not produce any additional phase in the material, although, the separation of the crystallographic planes reduced, indicating the presence of the Ti atoms in the crystal structure.

KEYWORDS: WS₂ nanosheet, Ti-dopant, thermal lens, thermal diffusivity.

I. INTRODUCTION

Two-dimensional (2D) semiconductors possess new properties in comparison to their bulk state [1]. Their 2D structures make them suitable materials for sensing and catalysis applications owing to their larger surface-area-to-volume ratio. Transition metal dichalcogenides (TMD) have been widely studied due to their unique and interesting properties like direct bandgap, water resistance, photoluminescence, thermoelectric properties, and good heat dissipation capability [2]–[4]. They consist of a transition metal layer sandwiched between

chalcogen (S, Se, or Te) layers [5]. In 2D configuration, many of TMDs show a transition from an indirect bandgap of the bulk state to a direct bandgap of the 2D materials [2]. Because of this new electronic structure, they can have high photoluminescence and also have higher carrier mobility [6]. With these properties, 2D TMDs are good candidates for LED and field-effect transistors (FET) [7]. 2D TMDs can be developed through exfoliation (mechanical or chemical methods) [8] of bulk TMDs or directly from their raw material by the chemical vapor deposition method (CVD). The last method has the advantage of generating isolated and high-quality 2D samples in cost-effective way [9]–[11].

The two dimensional tungsten disulfide (2D-WS₂) has a strong spin-orbital coupling [12], a direct bandgap of 2 eV, high charge mobility, and high thermal stability [6]. It is a good candidate for many applications like hydrogen generation, sensing of humidity [13], sensing different gases like NO₂ and H₂S [14], fluorescence enhancement [15], photodetector [16], LED or a variety of electronic devices [6], [12]. For such applications especially in microelectronic devices, heat management plays a vital role and thermal characterization of the constituent components is very important.

The electrical and optical properties of 2D TMDs are studied extensively [5], [17], [18],

but there are scarce experimental reports about their thermal parameters. The thermal conductivity (κ) value depends on the sample's geometry and for the same compound structure, different values are reported depending on the size and the thickness of the 2D films [4]. Most of the reports on κ measurement are done using Raman spectroscopy or thermoreflectance method [12]. In the first method, the shift in Raman peaks under laser illumination is recorded, and by using the data from Raman technique and introducing thermal modeling, κ of 2D TMD samples is extracted [2]. The errors of this technique arise from uncertainty in measuring the heat input as well as the temperature and effects of thermomechanical stress which could affect Raman bands [19]. In thermoreflectance technique, a modulated high-power exciting laser induces periodical temperature oscillations on the sample surface that will affect the reflection of the probe laser. Then, κ can be evaluated using the intensity and phase of the reflected probe beam [20]. One important drawback of this technique is the need for a smooth surface which limits its applications [21]. Photothermal lens technique, also has been used to evaluate D and κ of thin films [18,19].

Photothermal lens technique is a non-contact optical method and a promising way to measure thermal parameters of transparent or semi-transparent samples [22]–[24]. In the pump-probe photothermal lens configuration, an excitation laser is used to excite the sample and create a temperature gradient around the laser beam axis. As a result, the temperature gradient produces a refractive index gradient with an azimuthal symmetry (assuming that the excitation laser beam profile is Gaussian). Then the excited region in the sample acts like a lens named thermal lens (TL). When the probing laser beam passes through the sample, its phase is affected by the TL effect leads to a divergent or convergent beam depending on the nature of the material [25]–[27]. By using a detector and measuring the variation of the probe beam intensity at the detector position, the photothermal lens signal can be recorded. Using the temporal evolution of the TL signal, thermal diffusivity and thermal conductivity

can be determined [28], [29]. One important feature of TL method is a direct evaluation of the thermal diffusivity D which is an important property when studying unsteady-state heat transfer.

In this study, 2D titanium doped tungsten disulfide (Ti-doped WS₂) was prepared by CVD method. WS₂ has directly deposited on fluorine doped tin oxide (FTO) transparent substrates and arranged to vertical nanosheets. Despite an indirect bandgap for its bulk state, it possesses a direct bandgap of 2 eV. Ti atomic radius is close to that of W, and Ti can substitute W in WS₂ structure forming covalence bonds with S atoms in the lattice. The first report of doping WS₂ by Ti was done by our group and it shows an enhancement in electronic and optical properties such as photocurrent density compared to pure WS₂ and Mo-doped WS₂ [6].

In this work, we investigate the thermo-optical properties of 2D Ti-doped WS₂ as a function of Ti-dopant concentration. Moreover, the morphology and structural parameters were obtained by field emission scanning electron microscopy (FESEM), UV-Vis spectrophotometry, X-ray diffraction (XRD), and energy dispersive scanning (EDS) as well as Raman Spectroscopy.

II. THEORETICAL PART

In the TL model approach, developed by Shen *et al.* the TL signal $S(z, t)$ is defined as the relative change of the light transmission $T(z, t)$ through a small aperture when the excitation beam impinges on the sample [30]:

$$S(z, t) = \frac{T(z, t) - T(0)}{T(0)} \quad (1)$$

$T(0)$ is the transmittance when the excitation beam is off. According to this model, the TL signal can be expressed as a function of the photothermal and geometrical parameters of the sample as follows:

$$S(z, t) = \Phi_0 \tan^{-1} \left\{ \frac{4m(z)v(z)t/t_c(z)}{[v^2(z) + [1 + 2m(z)]^2 + [1 + 2m(z) + v^2(z)]^2 2t/t_c(z)]} \right\} \quad (2)$$

$$\Phi_0 = \frac{P_e \alpha l}{\lambda_p \kappa} \frac{ds}{dT} \quad (3)$$

$$m(z) = \frac{w_p^2(z)}{w_e^2(z)} \quad (4)$$

$$w_{p,e}(z) = w_{0p,0e} \sqrt{1 + \left(\frac{z - a_{p,e}}{z_{p,e}} \right)^2} \quad (5)$$

$$w_{0p,0e} = \sqrt{z_{p,e} \frac{\lambda_{p,e}}{\pi}} \quad (6)$$

$$v(z) = \frac{z - a_p}{z_p} + \frac{z_p}{L - z} \left[1 + \left(\frac{z - a_p}{z_p} \right)^2 \right] \quad (7)$$

where, $\lambda_{p,e}$, $w_{p,e}(z)$, $w_{0p,0e}$, $a_{p,e}$ and $z_{p,e}$ are the wavelength, beam radii, beam waists, position of the beam waists, and Rayleigh parameters of probe and excitation beams, respectively. P_e , α , l , ds/dT , κ , t_c , L are, the power of the excitation laser beam, absorption coefficient, sample's thickness, the temperature dependence of the optical path change, thermal conductivity, characteristic thermal lens time constant, and the distance between the sample (which is placed at $z = 0$) and the detector [31].

III. EXPERIMENTAL PART

A. Samples preparation

The substrates are 2 mm thickness glass coated with ~500 nm FTO (Dyesol Ltd.). FTO as a transparent (visible transmittance~80%) and conducting (conductivity~ $3 \times 10^{-4} \Omega^{-1} \text{cm}^{-1}$) substrate was also used as an electrode for a further characteristics of the samples [32]. Vertical WS₂ nanosheets were produced by chemical vapor deposition (CVD) method using a mixture of argon/hydrogen atmosphere at a pressure of 10^{-2} Torr and flow rate of 40

sccm in a tube furnace. TiO₂, WO₃ (99.7%), and S powder (99.5%) were purchased from α -Aeser and used as precursors for Ti, W, and S. Different dopant concentrations are prepared by using different TiO₂ to WO₃ powders mixture in the CVD furnace. The used powders ratios are listed in Table 1. The position for TiO₂, WO₃, and S powders is defined by the maximum temperature of the points in the furnace. For TiO₂, WO₃ was at the center ($T_{\text{max}} \sim 500^\circ\text{C}$) where also the substrates are placed and for S powder at the flow entrance of the furnace ($T_{\text{max}} \sim 180^\circ\text{C}$). Increment of the furnace temperature for the sample deposition was done in 20 min from room temperature to 500°C , after reaching the growth temperature at 500°C stabilized for 3 min. The quartz tube at the end of the process is naturally cooled down to room temperature.

Table 1. Different powders mixture used to introduce different Ti-doping concentrations

TiO ₂ /WO ₃ powders ratio	Sample name
0/100	pure WS ₂
5/95	7% Ti-doped WS ₂
10/90	14% Ti-doped WS ₂
20/80	28% Ti-doped WS ₂

B. Photothermal Lens Experimental Setup

The experimental setup is schematically shown in Fig. 1. A diode-pumped 532 nm solid-state laser (MGL-III-532-100, UltraLasers, Canada) is used as an excitation laser (EL).

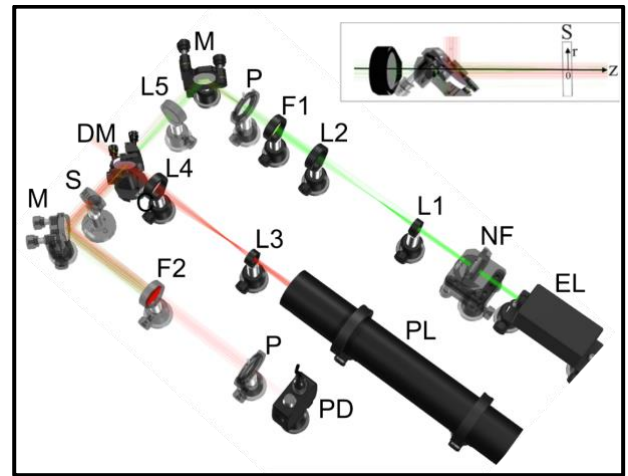


Fig. 1. Schematic of the pump-probe photothermal lens setup: EL: excitation laser, PL: probe laser, PD: photodiode, NF: neutral density filter, L1, L2, L3, L4, L5: lens, F1 and F2 band-pass filters, M: mirror, DM: dichroic mirror, P: pinhole, S: sample, the inset

shows the configuration regarding the z-axis and the radial direction (r) of the optical configuration.

A signal generator (RIGOL DG1022, RIGOL Technologies, Inc, USA) modulates the EL at 3 Hz. A neutral density filter (NF) (Thorlabs, NDC 50S-3) is placed after the laser to adjust the power of the EL to 24 mW on the sample. The EL is first collimated by the lenses L1 (LB1757-A, Thorlabs, USA) and L2 (LB1676-A, Thorlabs, USA) and then filtered its first harmonic 1064 nm by filter F1 (Thorlabs, NF533-17, center wavelength 532 nm). After that, the beam passes through the lens L5 (LB1437-A, Thorlabs, USA) and is focused to a 40 μm diameter spot into the sample (S). Consequently, the Rayleigh distance (3 mm) is larger than the sample length to guarantee a constant radius of the beam over the sample path length and the gradient of the heat flow to be mainly in the radial direction. The probe laser (PL), a 5 mW He-Ne laser (05-UR-111, Melles Griot, USA) is collimated by a set of lenses L3 (LB1757-A, Thorlabs, USA) and L4 (LB1676-A, Thorlabs, USA) and directed to the sample collinearly with the EL. After interacting with the sample, the PL's intensity changes are detected by the photodiode (PD) (PDA 36A-EC, Thorlabs, USA) with a 0.5 mm pinhole (P) and a 632 nm filter F2 (MELLES GRIOT, USA). The signal from the PD is measured and monitored by the oscilloscope (RIGOL DS1102E, RIGOL Technologies, Inc, USA). The inset in Fig. 1 shows the configuration regarding the excitation and probe beams propagating collinearly in the z-axis direction.

IV. RESULTS AND DISCUSSION

A. Absorbance and Structural Characterization

The UV-Visible absorption spectra shown in Fig. 2 were measured using a Uv-Vis spectrophotometer (Pishro Pazhohesh, IR-Iran). Absorption spectra show the main absorption peaks at wavelengths: 420, 480, 540, and 650 nm. The UV-Visible absorption results indicate that higher Ti-dopant concentrations resulted in higher absorbance in the wavelength related to the excitation laser (532nm). One

explanation could be that Ti-dopant tuned the structure bandgap results in absorbance enhancement.

Fig. shows the FESEM image of WS₂ nanosheets, which indicates their alignment vertical to the substrate. The same geometrical characteristics are seen for all samples regardless of their different doping concentrations. The surface area of the nanosheets varies from around 200 nm to 2 μm in width.

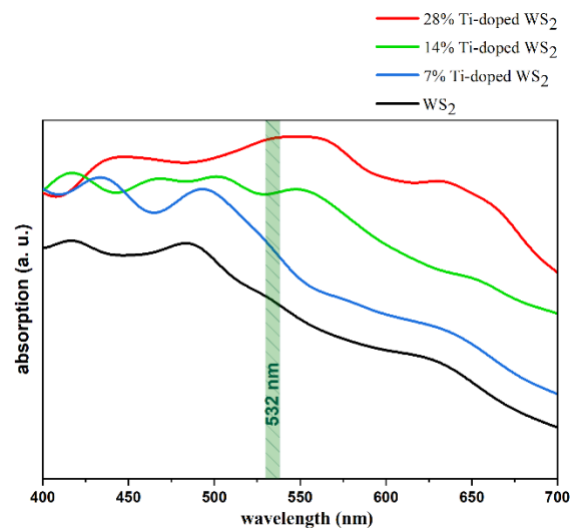


Fig. 2. UV-Visible absorption spectra for Ti-doped WS₂ samples at different concentrations. The green crosshatch indicate the region related to the excitation laser wavelength.

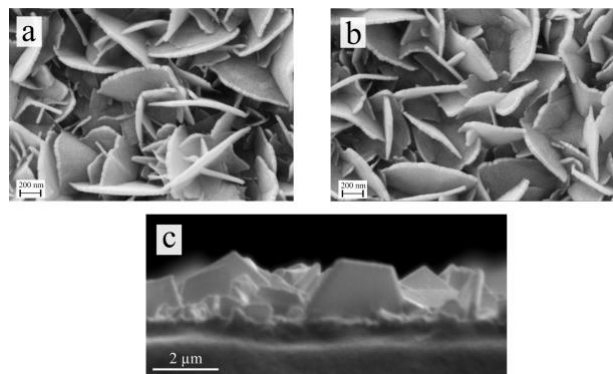


Fig. 3. FESEM image of (a) pure WS₂ and (b) 28% Ti-doped WS₂ vertical nanosheets, and (c) a side view of the 28% WS₂ vertical nanosheets.

Fig. shows the XRD pattern using Cu K α radiation (Philips Netherlands). The results show that the main peak of the WS₂ crystal structure varies between 14 and 15 degrees (circled in Fig. 4), which corresponds to (002) crystal planes (JCPDS Card Number 98-003-

9096). The rest of the peaks correspond to FTO substrate. The XRD analysis (Fig. a) of pure WS₂, (7%, 14% and 28% Ti-doped WS₂ samples) indicates no evidence of TiO₂ structure. Additionally, for pure WS₂ (Fig. b), the peak (002) is located at 14.23° and shifts to higher degrees 14.31°, 14.39°, and 14.41° for 7%, 14%, and 28% Ti-doped WS₂ samples, respectively. According to Bragg's equation, this shift indicates that the d-spacing between (002) crystallographic planes decreases with increasing the Ti-dopant concentrations. The reason for this shift could be the substitution of W with Ti in the crystal structure of WS₂ [6].

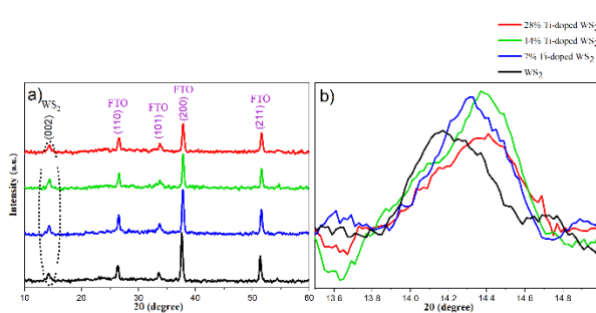


Fig. 4. XRD pattern of (a) pure, 7%, 14%, and 28% Ti-doped WS₂ nanosheets. The circled part corresponds to WS₂ peak and other peaks are related to FTO substrate, (b) characteristic XRD (002) peak of WS₂ crystal structure for all samples in (a).

Ionic radii for both Ti and W are close to each other but the distortion that happens to the lattice as a result of dopants causes a contraction which results in an increment of the related XRD peak [33].

To analyze the constituent elements of the samples, energy dispersive X-ray spectroscopy (EDS) has been used. Figure 5 shows the EDS for 7% and 28% Ti-doped WS₂ samples.

The abundances of elements in the samples are in order S, W, and then Ti as it is expected from the abundances in the composition. The defined dopant percentage in this work is calculated by the output of the EDS analysis as the ratio of Ti to W.

Raman spectroscopy analyses of the samples are shown in Fig. 6(a). The main two peaks are related to in-plane (E_{2g}^1) and out-of-plane (A_{1g}) vibrations of the sulfur atoms in the crystal structure [34]. To evaluate more accurately the

information related to A_{1g} peak, multiple peak fit analysis has been used (Fig. 6(b)). The A_{1g} peak for the 28% Ti-doped WS₂ sample shifted (4 cm^{-1}) to higher frequencies. This result agrees with the XRD analysis which also indicates a decrease in the spacing between the (002) planes (Fig.4b). The decrease in the spacing of the out-of-plane or normal direction to the WS₂ sheets could be used as a confirmation for the substitution of titanium in the crystal structure [35]. The peaks between 500 and 700 cm^{-1} correspond to higher order Raman modes [36].

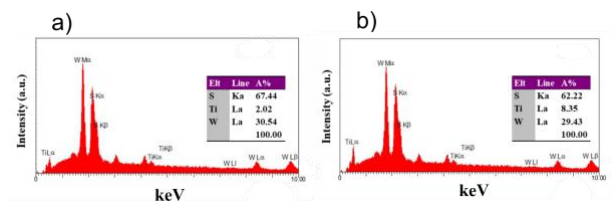


Fig. 5. EDS analysis for (a) 7% and (b) 28% Ti-doped WS₂ nanosheets.

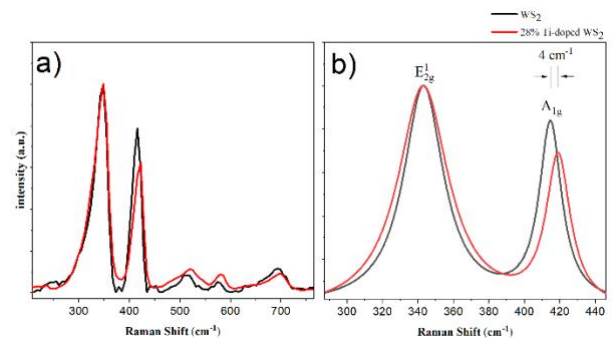


Fig. 6. (a) Raman spectra of the pure WS₂ and 28% Ti-doped WS₂ nanosheets structure. (b) the main two characteristic Raman peaks for WS₂ crystal structure with the 4 cm^{-1} shift by using peak fit in origin software.

B. Photothermal lens results

For calibrating the setup, D of ethanol was measured, and the value ($0.94 \times 10^{-1}\text{ mm}^2/\text{s}$) agrees well with the literature data [37]. Subsequently, we proceed to measure Ti-doped WS₂ samples. The results show that increasing Ti-doping produces a higher TL signal (Fig. 7), in agreement with the UV-Vis measurement (Fig. 2).

Parameter D of the samples is evaluated by the best fit to the TL experimental signals using Eq.2 for known experimental parameters such as the distance between the detector and the

sample, $L=28.5$ cm and the Rayleigh distances, $z_e=0.3$ cm, $z_p=10000$ cm. Considering that the Rayleigh distance of the excitation beam is much larger than the thickness of the examined sample, the heat transfer through the sample will be mainly in the perpendicular direction to the laser beam axis and therefore, parallel to the substrate (in-plane). Note that, the value of the absorbance for the glass substrate is relatively very low in comparison to the layer of WS₂, therefore, the measured D is mainly because of the in-plane transmission of heat. Calculated D and estimated κ using the mentioned approach are given in Table 2. For calculating κ , we have used a very-well known dependence of D as reported in the reference [38].

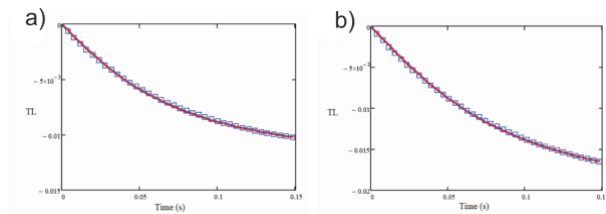


Fig. 7. TL signal as a function of time and the best-fitted curve for (a) 7% and (b) 28% Ti-doped WS₂ samples at 3 Hz modulation frequency. The average RSD for 3 measured values lies in the range of 8%-10%.

We could find only κ but not D of WS₂ nanostructures in previous studies. The values for D and κ of 2D TMDs and also bulk W reported in the literature are listed in Table 3. The reported values for κ of 2D WS₂ are in the range of 1 to 50 W/m.K depend on the fabrication method and the size of the samples [12],[39]. In this study, the TL signal as well as D increased with Ti-dopant concentration in WS₂ nanosheets. In reference [6], the evaluation of optical bandgap and photocurrent density for both the pure WS₂ sample and 28% Ti-doped WS₂ gives respectively, 1.93 and 1.69 eV for the bandgap and 32 and 175 $\mu\text{A}/\text{cm}^2$ for the photocurrent density under illumination power of 100 mW/cm². Therefore, in Table 2 the increment in thermal diffusivity could be related to the lower bandgap and higher carrier mobility of the Ti-doped WS₂ samples that have a role in the sample thermal behavior as well.

Table 2 Thermal parameters for WS₂ and Ti-doped WS₂

Sample name	$D(10^{-6} \text{ m}^2/\text{s})$	$\kappa(\text{W}/\text{m.K})$
pure WS ₂	0.24 ± 0.03	$0.7 >$
7% Ti-doped WS ₂	0.5 ± 0.04	1
14% Ti-doped WS ₂	0.8 ± 0.07	2
28% Ti-doped WS ₂	1.3 ± 0.1	4

Table 3 Reported thermal conductivity and thermal diffusivity in some literature

Sample	method	κ (W/m.K)	D (mm ² /s)	Ref.
exfoliated single layer MoS ₂	Raman spectroscopy	35	-	[38]
exfoliated few layers MoS ₂	suspended micro-devices	44-52	-	[38]
exfoliated few layers MoS ₂	Raman spectroscopy	52	-	[38]
single layer MoS ₂	Raman spectroscopy	84 ± 17	-	[40]
single layer MoSe ₂	Raman spectroscopy	59 ± 18	-	[40]
Bulk MoS ₂	Time-resolved magneto-optic Kerr effect	85-112	-	[38]
single layer WS ₂	Raman spectroscopy	32	-	[12]
single layer WS ₂	Raman spectroscopy	15, 28	-	[41,42]
Double layer WS ₂	Raman spectroscopy	53	-	[12]
single layer MoS ₂	Raman spectroscopy	34.5	-	[43]
Bulk W		135	27	[44]

V. CONCLUSION

Ti-doped WS₂ perpendicular nanosheets have been fabricated on fluorine doped tin oxide (FTO) substrates using a quasi-low temperature chemical vapor deposition (CVD) technique. Thermal lens (TL) signal measurement was

used to evaluate Thermal diffusivity (D) ($0.24-1.3 \text{ mm}^2/\text{s}$) and thermal conductivity, κ ($\leq 4 \text{ W/m.K}$) of Ti-doped WS_2 nanosheets. Higher dopant concentrations possess higher absorption, TL signal, and D Ti-dopants influence WS_2 crystal features accompanied by reduction of two dimensional (2D) layers spacing that evidence by a shift in the XRD and also Raman peaks. The results demonstrate the ability of the mode-mismatched photothermal lens technique to calculate D in delicate thin films. These measurements are important for photonic devices, to design the proper heat removal system which is critical to avoiding thermal lensing and other deleterious effects on the device application induced by the temperature gradient.

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